



US006143655A

# United States Patent [19]

[11] Patent Number: **6,143,655**

Forbes et al.

[45] Date of Patent: **Nov. 7, 2000**

[54] **METHODS AND STRUCTURES FOR SILVER INTERCONNECTIONS IN INTEGRATED CIRCUITS**

[75] Inventors: **Leonard Forbes**, Corvallis, Oreg.; **Paul A. Farrar**, So. Burlington, Vt.; **Kie Y. Ahn**, Chappaqua, N.Y.

[73] Assignee: **Micron Technology, Inc.**, Boise, Id.

[21] Appl. No.: **09/030,113**

[22] Filed: **Feb. 25, 1998**

[51] Int. Cl.<sup>7</sup> ..... **H01L 21/00**; H01L 21/3205; H01L 21/44; H01L 29/40

[52] U.S. Cl. .... **438/686**; 438/46; 438/85; 438/602; 438/660; 438/752; 438/770; 257/742; 257/743; 257/762

[58] Field of Search ..... 438/602, 686, 438/660, 658, 85, 46, 752, 770, 779, 796; 257/762, 743, 742

[56] **References Cited**

**U.S. PATENT DOCUMENTS**

4,702,941	10/1987	Mitchell et al. ....	427/250
4,855,252	8/1989	Peterman et al. ....	437/189
4,959,705	9/1990	Lemnius ..... ..	257/522
5,148,260	9/1992	Inoue et al. .... ..	257/276
5,187,560	2/1993	Yoshida et al. ....	257/743
5,324,684	6/1994	Kermani et al. ....	437/95
5,341,016	8/1994	Prall et al. .... ..	257/412
5,371,035	12/1994	Pfiester et al. ....	437/69
5,470,801	11/1995	Kapoor et al. .... .	437/238
5,510,645	4/1996	Fitch et al. .... ..	257/522
5,563,448	10/1996	Lee et al. .... .. .	257/742
5,710,454	1/1998	Wu ..... .. .	257/413
5,757,072	5/1998	Gorowitz et al. ....	257/522
5,796,151	8/1998	Hsu et al. .... .. .	257/410
5,796,166	8/1998	Agnello et al. ....	257/751
5,801,444	9/1998	Aboelfotah et al. .	257/742
5,925,918	7/1999	Wu et al. .... .. .	257/413
5,942,799	8/1999	Danek et al. .... .	257/751

**OTHER PUBLICATIONS**

Berezhnoi, A., *Silicon and its Binary Systems*, Consultants Bureau, New York, 84, (1960).

Fukuda, Y., et al., "A New Fusible-Type Programmable Element Composed of Aluminum and Polysilicon", *IEEE Trans. on Electron Devices*, ED-33, 250-253, (Feb., 1986).

Hanna, J., et al., "Early Stage of Polycrystalline Growth of Ge and SiGe by Reactive Thermal CVD from GeF(4) and Si(2)H(6)", *Materials Res. Soc. Symp. Proc.*, 358, Boston, MA, 877-881, (Nov./Dec., 1994).

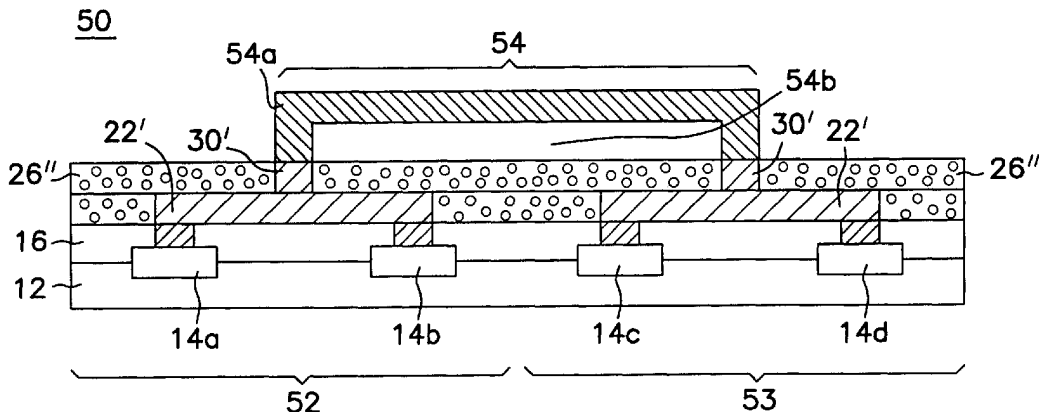
(List continued on next page.)

*Primary Examiner*—John F. Niebling  
*Assistant Examiner*—Ha Tran Nguyen  
*Attorney, Agent, or Firm*—Schwegman, Lundberg, Woessner & Kluth, P.A.; Eduardo E. Drake

[57] **ABSTRACT**

A typical integrated-circuit fabrication requires interconnecting millions of microscopic transistors and resistors with aluminum wires. Making the aluminum wires flush, or coplanar, with underlying insulation requires digging trenches in the insulation, and then filling the trenches with aluminum to form the aluminum wires. Trench digging is time consuming and costly. Moreover, aluminum has higher electrical resistance than other metals, such as silver. Accordingly, the invention provides a new "self-trenching" or "self-planarizing" method of making coplanar silver wires. Specifically, one embodiment forms a first layer that includes silicon and germanium; oxidizes a region of the first layer to define an oxidized region and a non-oxidized region; and reacts silver with the non-oxidized region. The reaction substitutes, or replaces, the non-oxidized region with silver to form silver wires coplanar with the first layer. Another step removes germanium oxide from the oxidized region to form a porous insulation having a very low dielectric constant, thereby reducing capacitance. Thus, the present invention not only eliminates the timing-consuming, trench-digging step of conventional methods, but also reduces resistance and capacitance which, in turn, enable faster, more-efficient integrated circuits.

**28 Claims, 3 Drawing Sheets**



## OTHER PUBLICATIONS

- Hansen, P., *Constitution of Binary Alloys*, McGraw-Hill, New York, 103, (1958).
- Hiraki, A., et al., "Formation of Silicon Oxide over Gold Layers on Silicon Substrates", *J. Applied Physics*, 43, 3643-3649, (Sep., 1972).
- Hiraki, A., et al., "Low-Temperature Migration of Silicon in Metal Films on Silicon Substrates Studied by Backscattering Techniques", *J. Vacuum Science and Tech.*, 9, 155-158, (Jan./Feb., 1972).
- Horie, H., et al., "Novel High Aspect Ratio Aluminum Plug for Logic/DRAM LSI's Using Polysilicon-Aluminum Substitute", *Technical Digest: IEEE Int. Electron Devices Meeting*, San Francisco, CA, 946-948, (1996).
- Hurley, P., et al., "Low Temperature Plasma Oxidation of Polycrystalline Silicon", *Proc. 7th European Conf. on Insulating Films on Semiconductors: Contributed Papers, Section 5*, IOP Publishing Ltd., 235-238, (1991).
- Hwang, J., et al., "Novel Polysilicon/TiN Stacked-Gate Structure for Fully-Depleted SOI/CMOS", *Technical Digest: Int. Electron Devices Meeting*, San Francisco, CA, 345-348, (Dec. 1992).
- King, T., et al., "Deposition and Properties of Low-Pressure Chemical-Vapor Deposited Polycrystalline Silicon-Germanium Films", *J. Electrochemical Society*, 141, 2235-2241, (Aug. 1994).
- Lee, D.H., et al., "Gate Oxide Integrity (GOI) of MOS transistors with W/TiN stacked gate", *1996 Symposium on VLSI Technology Digest of Technical Papers*, 208-209, (1996).
- Li, C., et al., "Low Temperature Heteroepitaxial Growth of Si(1-x)Ge(x)-on-Si by Photo-Enhanced Ultra High Vacuum Chemical Vapor Deposition Using Si(2)H(6) and Ge(2)H(6)", *J. Electronic Materials*, 24, 875-884, (Jul. 1995).
- Li, P., et al., "Formation of Stoichiometric SiGe Oxide by Electron Cyclotron Resonance Plasma", *Appl. Phys. Lett.*, 60, 3265-3267, (Jun. 1992).
- Lyman, T.e., "Metallography, Structure and Phase Diagrams", *Metals Handbook*, 8, American Society for Metals; Metals Park, Ohio, 253, 256, 260, 263., (1989).
- Moffatt, W., *The Handbook of Binary Phase Diagrams*, General Electric Company, pub., vol. 1, 3/84, (1978).
- Mohajerzadeh, S., et al., "A Low Energy Ion Beam Assisted Deposition Technique for Realizing iso-type SiGe/Si hetero-interface diodes", *Thin Solid Films*, 283, 182-187, (1996).
- Mohajerzadeh, S., et al., "A Low-Temperature Ion Vapor Deposition Technique for Silicon and Silicon-Germanium Epitaxy", *Canadian J. Physics*, 74, S69-S73, (1996).
- Mohri, M., et al., "Effect of SiF(4)/SiH(4)/H(2) Flow Rates on Film Properties of Low-Temperature Polycrystalline Silicon Films Prepared by Plasma Enhanced Chemical Vapor Deposition", *IEICE Transactions on Electronics*, E77-C, 1677-1684, (Oct. 1994).
- Mukhopadhyay, M., et al., "Properties of SiGe Oxides Grown in a Microwave Oxygen Plasma", *J. Applied Physics*, 78, 6135-6140, (Nov. 1995).
- Predel, B., et al., "Die Zustandsdiagramme Silber-Germanium-Silizium und Gold-Germanium-Silizium", *J. Less-Common Metals*, 44, 39-49, (Jan. 1976).
- Schadel, H., et al., "Activity of Liquid Silver-Silicon Alloys", *Trans. American Institute of Mining and Metallurgical Engineers*, 188, 1282-1283, (Oct. 1950).
- Ushiku, Y., et al., "Planarized Silver Interconnect Technology with a Ti Self-Passivation Technique for Deep Sub-Micron ULSIs", *1993 Symp. on VLSI Technology: Digest of Technical Papers*, 121-122, (1993).
- Wu, S., et al., "Suppression of the Boron Penetration Induced Si/SiO<sub>2</sub> Interface Degradation by Using a Stacked-Amorphous-Silicon Film as the Gate Structure for pMOSFET", *IEEE Electron Device Letters*, 15, 160-162, (May 1994).
- Cherkasov, Y.A., et al., "Optical information media based on an amorphous photoconductor with high avalanche amplification—the amorphous analog of trigonal selenium", *Journal of Optical Technology*, vol. 64, No. 7, 627-32, (Jul. 7, 1997).
- Shuto, S., et al., "Read Disturb Degradation Mechanism for Source Erase Flash Memories", *1996 Symposium on VLSI Technology, Digest of Technical Papers*, Honolulu, HI, 242-43, (1996).

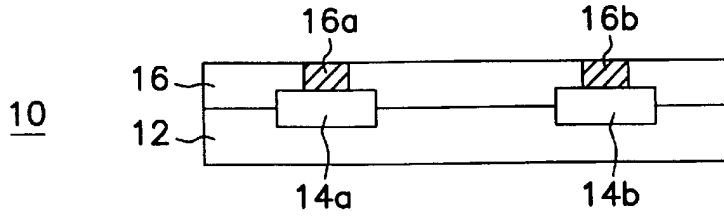


FIG. 1

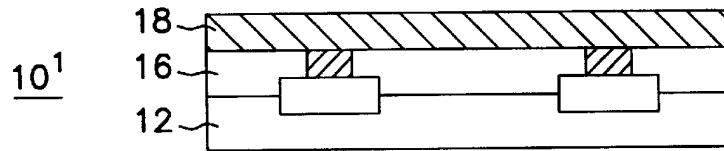


FIG. 2

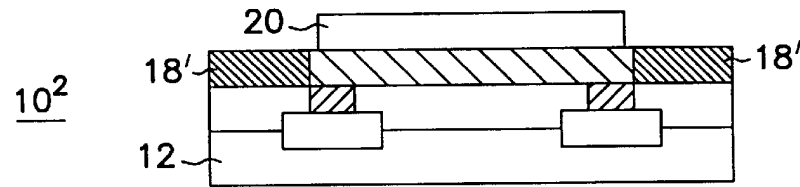


FIG. 3A

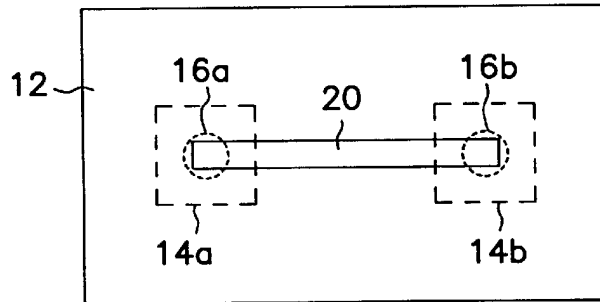


FIG. 3B

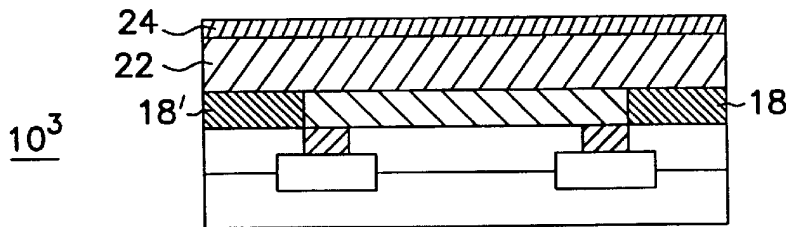


FIG. 4

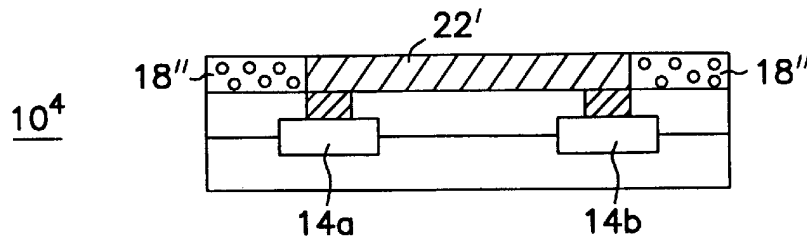


FIG. 5

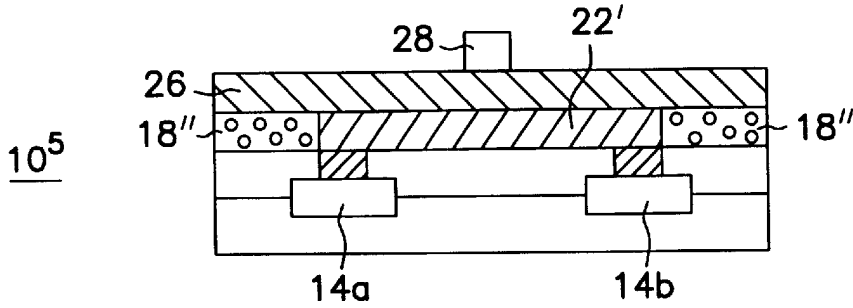


FIG. 6A

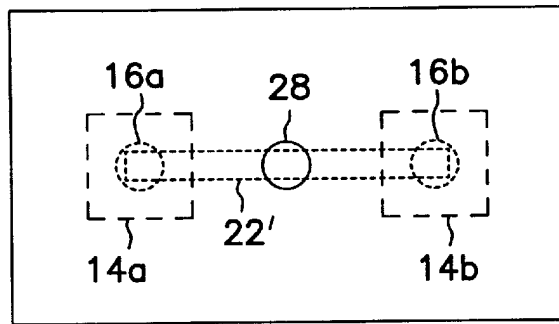


FIG. 6B

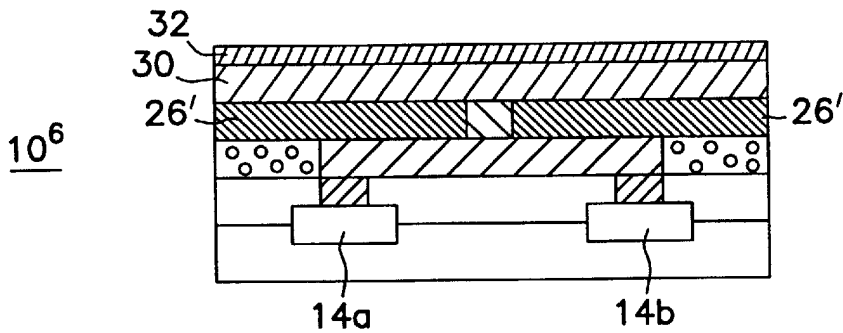


FIG. 7

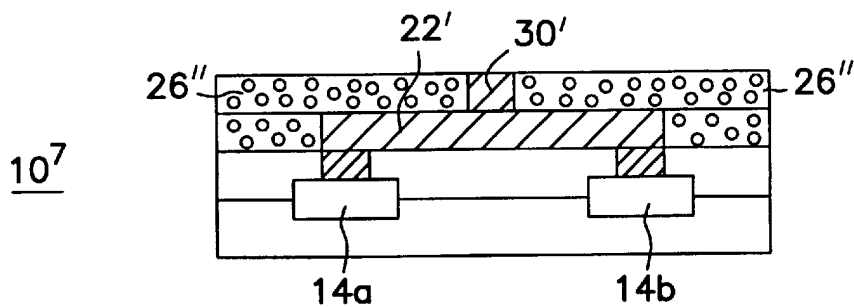


FIG. 8

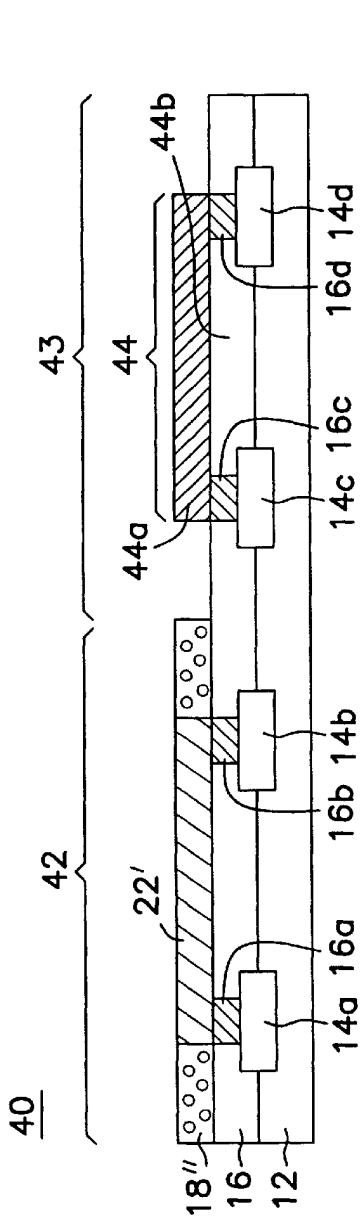


FIG. 9

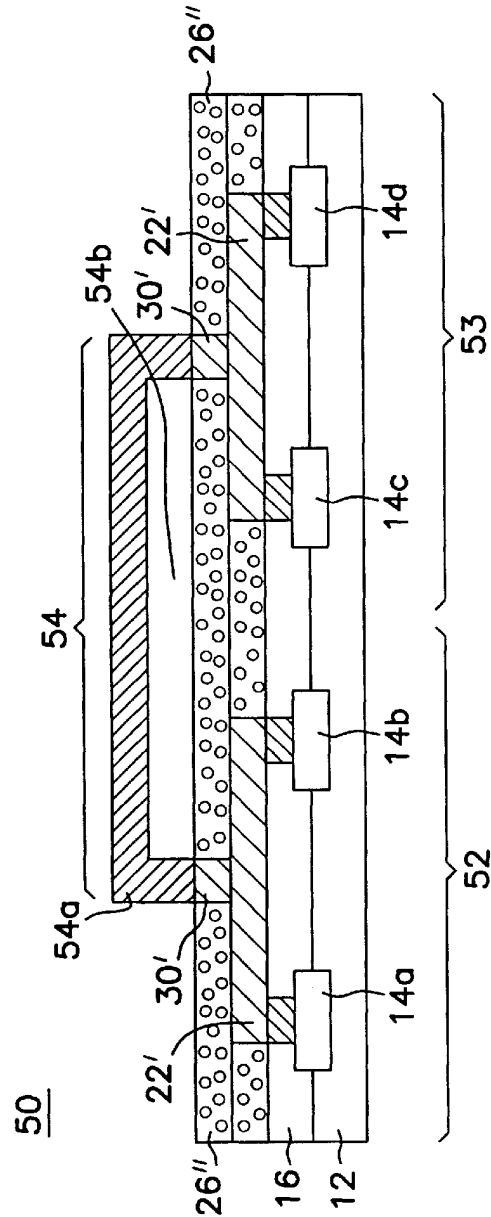


FIG. 10

## METHODS AND STRUCTURES FOR SILVER INTERCONNECTIONS IN INTEGRATED CIRCUITS

### BACKGROUND OF THE INVENTION

The present invention concerns methods of making, or fabricating, integrated circuits, particularly methods of forming silver interconnects.

Integrated circuits, the key components in thousands of electronic and computer products, are interconnected networks of electrical components fabricated on a common foundation, or substrate. Fabricators typically use various techniques, such as layering, doping, masking, and etching, to build thousands and even millions of microscopic resistors, transistors, and other electrical components on a silicon substrate, known as a wafer. The components are then "wired," or interconnected, together to define a specific electric circuit, such as a computer memory or microprocessor.

Interconnecting millions of microscopic components typically entails covering the components with an insulative layer, digging small holes in the insulative layer to expose portions of the components underneath, and digging trenches from each hole to one or more other holes in the layer. Then, through metallization, the holes and trenches are filled with aluminum (or an aluminum alloy) to form aluminum interconnects, or wires, between the components.

To fill the trenches and holes, fabricators cover the entire insulative layer with a thin layer, or film, of aluminum, and then selectively dissolve, or etch, away the aluminum that lies outside the holes and trenches. The selective etching requires the use of photolithography, a photographic-patterning technique, to form an etch-resistant mask, which protects the aluminum-filled holes and trenches from the etchant. The resulting aluminum wires, intended to be flush, or coplanar, with the surface of the underlying insulative layer, are typically about one micron thick, or about 100 times thinner than a human hair.

These conventional interconnection techniques suffer from at least three significant shortcomings. First, because of the difficulty of using photolithography to form high-precision masks on bumpy, uneven surfaces, conventional techniques require digging trenches to ensure that the deposited aluminum wires are flush, or coplanar, with the surface of the underlying insulation. However, digging these trenches is a time-consuming step which ultimately increases the cost of manufacturing integrated circuits.

Secondly, conventional techniques produce wires of aluminum, which not only has a higher electrical resistance, but also a lower electromigration resistance than other metals, such as silver. High electrical resistance wastes power, and low electromigration resistance means that, at certain electric current levels, the aluminum readily diffuses, or migrates, into neighboring structures, eventually thinning or breaking the wires and thus undermining reliability of integrated circuits.

Moreover, although silver has a 40-percent lower electrical resistivity and at least 100-percent higher electromigration resistance than aluminum, conventional interconnection techniques are impractical for making silver interconnects. In particular, silver, a noble metal, is immune to most etchants. In fact, attempts to selectively etch a layer of silver covered with an etch-resistant mask usually dissolve the mask, not the silver. Thus, conventional etch-based techniques are wholly inadequate to form silver interconnects.

Thirdly, in addition to being time-consuming because of the trench-digging step and ineffective with more desirable

metals such as silver, conventional techniques place aluminum wires in relatively high-capacitance insulators, typically solid silicon oxide. High capacitance slows the response of integrated circuits to electrical signals, a great disadvantage in computers and other systems including the integrated circuits.

Accordingly, there is not only a need for new interconnection methods that eliminate the trench-digging step, but also methods that yield less-resistive, less-capacitive, and more-reliable silver-based interconnects for faster and more-efficient integrated circuits.

### SUMMARY OF THE INVENTION

To address these and other needs, the present invention provides a new method of making coplanar silver and insulative structures for an integrated circuit. Specifically, one embodiment of the method entails forming a first layer that includes silicon and germanium, and then oxidizing a region of the first layer to define an oxidized region and a non-oxidized region. After oxidation, the method reacts silver with the non-oxidized region. The reaction substitutes, or replaces, the non-oxidized region with silver to form a silver structure flush or coplanar with the first layer. Another step removes germanium oxide from the oxidized region to surround the silver structure in a porous insulative member which reduces capacitance.

Thus, the method of the present invention yields a self-planarizing silver structure that not only eliminates the time-consuming, trench-digging step of conventional methods, but also places the low-resistance, highly-reliable silver structure within a capacitance-reducing insulation that allows faster, more-efficient integrated circuits.

### BRIEF DESCRIPTION OF THE DRAWINGS

The following figures are used to describe many aspects of the invention:

FIG. 1 is a cross-sectional view of an integrated-circuit assembly;

FIG. 2 is a cross-sectional view of the FIG. 1 integrated-circuit assembly after formation of a silicon-germanium layer;

FIG. 3A is a cross-sectional view of the FIG. 2 integrated-circuit assembly after formation of an oxidation mask;

FIG. 3B is a top view of the FIG. 3A integrated-circuit assembly, showing the oxidation mask;

FIG. 4 is a cross-sectional view of the FIG. 3 integrated-circuit assembly after oxidation, removal of the oxidation mask, and formation of a silver layer and a zirconium layer;

FIG. 5 is a cross-sectional view of the FIG. 4 assembly after reaction of the silver layer and the silicon-germanium layer.

FIG. 6A is a cross-sectional view of the FIG. 5 integrated-circuit assembly after formation of a silicon-germanium layer and an oxidation mask;

FIG. 6B is a top view of the FIG. 6A integrated-circuit assembly, showing the oxidation mask;

FIG. 7 is a cross-sectional view of the FIG. 6 integrated-circuit assembly after oxidation, removal of the oxidation mask, and formation of a silver layer and a titanium layer;

FIG. 8 is a cross-sectional view of the FIG. 7 assembly after reaction of the silver layer and the silicon-germanium layer;

FIG. 9 is a cross-sectional view of an integrated-circuit assembly embodying a coplanar hybrid interconnect system; and

FIG. 10 is a cross-sectional view of an integrated-circuit assembly embodying a two-level (non-coplanar) hybrid interconnect system.

#### DESCRIPTION OF THE PREFERRED EMBODIMENTS

The following detailed description, which references and incorporates FIGS. 1–10, describes and illustrates specific embodiments of the invention. These embodiments, offered not to limit but only to exemplify and teach the invention, are shown and described in sufficient detail to enable those skilled in the art to implement or practice the invention. Thus, where appropriate to avoid obscuring the invention, the description may omit certain information known to those of skill in the art.

The detailed description includes two sets of separate but overlapping embodiments of the invention. The first set of embodiments, illustrated principally with FIGS. 1–8, pertains to a preferred method of making single- and multi-level interconnective structures. The second set of embodiments, illustrated principally with FIGS. 9 and 10, concern hybrid interconnect structures which combine air bridges with structures exemplified in the first set of embodiments.

#### PREFERRED METHOD OF MAKING SINGLE- AND MULTI-LEVEL INTERCONNECTS

FIGS. 1–8 show a number of preferred integrated-circuit assemblies, which taken collectively and sequentially, illustrate the preferred method of making single- and multi-level interconnects. The method, as shown in FIG. 1, begins with a known integrated-circuit assembly or structure 10. Assembly 10 includes a substrate 12. The term “substrate” encompasses a semiconductor wafer as well as structures having one or more insulative, conductive, or semiconductive layers and materials. Thus, for example, the term embraces silicon-on-insulator, silicon-on-sapphire, and other advanced structures.

Substrate 12 supports a number of integrated elements 14, preferably transistors 14a and 14b. Transistors 14a and 14b are covered by an insulative layer 16, which preferably comprises silicon oxide, nitride, or oxynitride. Layer 16 includes two aluminum vias 16a and 16b electrically connected to respective transistors 14a and 14b. Although omitted from FIGS. 1–8 for clarity, assembly 10 preferably includes a titanium-nitride diffusion barrier between vias 16a and 16b and transistors 14a and 14b.

Next, as FIG. 2 shows, the preferred method forms a polycrystalline, silicon-germanium (SiGe) layer 18 on insulative layer 16 and over vias 16a and 16b. In the preferred embodiment, silicon-germanium layer 18 consists of 10–60 percent germanium (by weight). Because of the low solubility of silicon in silver, compared to silicon in germanium, as well as to produce a low-dielectric insulator, a high germanium content, such as 50 percent, is preferred. The thickness of the silicon-germanium layer, approximately one micron, matches the thickness of the desired silver structure.

Silicon-germanium layer 18 is preferably formed through conventional low-pressure chemical-vapor deposition (LP-CVD), specifically the pyrolysis of silicon hydride (SiH<sub>4</sub>) and germanium hydride (GeH<sub>4</sub>) gases at a temperature as low as 400° C. For more details on this technique, which controls the silicon-germanium composition by varying the gas flow rates, refer to the T.J. King article, entitled “Deposition and Properties of Low-Pressure Chemical-Vapor Deposited Polycrystalline Silicon-Germanium Films,” in the Journal of the Electro-Chemical Society (pp. 2235–41, August 1994). This article is incorporated by reference.

In FIG. 3A, the method forms an oxidation mask 20 on silicon-germanium layer 18. This mask defines a narrow, rectangular channel which ultimately defines the periphery of a conductor connecting vias 16a and 16b and thus transistors 14a and 14b. A top view of mask 20 and its relation to transistors 14a and 14b and vias 16a and 16b is shown in FIG. 3B.

To form mask 20, the method deposits a layer of silicon nitride (Si<sub>3</sub>N<sub>4</sub>), using low-pressure or plasma-enhanced chemical-vapor deposition. Conventional lithographic techniques define the mask, finally formed through reactive-ion or plasma etching of the silicon-nitride layer. The actual mask thickness, which preferably ranges between 30 and 100 nanometers, depends on the etchant used to remove the unwanted silicon nitride.

After completion of mask 20, the method oxidizes silicon-germanium layer 18. This oxidation defines regions under mask 20 as non-oxidized regions and unmasked regions as oxidized regions. The oxidized regions contain an extractable or removable germanium oxide, which can be removed to leave a porous insulator having a low dielectric constant. The non-oxidized regions eventually become the metallic or conductive member connecting transistors 14a and 14b. In the preferred embodiment, the oxidation is performed in microwave-enhanced plasma to occur at temperatures as low as 200° C. The results of oxidation are shown as regions 18' in FIG. 3A.

In FIG. 4, the method entails removing mask 20, preferably through reactive-ion etching, chemical etching, or chemical-mechanical polishing, to leave silicon-germanium layer 18 with a planar surface for depositing silver. Afterward, physical or chemical-vapor deposition forms a 1.5-micron-thick, silver layer 22 on the oxidized and non-oxidized regions of layer 18.

Next, the method forms a 200-nanometer-thick, zirconium layer 24 on layer 22 by physical vapor deposition. In other embodiments, layer 24 is between 20 and 250 nanometers thick and comprises titanium or hafnium, instead of zirconium. Zirconium is preferred because of its lower solubility in silver. Layer 24 not only reduces the temperature and time necessary to complete the next step, but also improves the contact resistance between the silver layer and any subsequent via level.

The next step forces a metal-substitution reaction between silver layer 22 and the non-oxidized regions of silicon-germanium layer 18, thereby substituting, or replacing, the non-oxidized regions of silicon-germanium layer 18 with silver from silver layer 22. Consequently, a coplanar (or at least partially submerged) metallic structure 22', consisting essentially of silver with small amounts of silicon and germanium, forms in silicon-germanium layer 18. Thus, unlike conventional techniques that require trench-digging before metal deposition to ensure co-planarity of the metallic structures and underlying insulative structures, the present method is “self-planarizing” or “self-trenching.”

More specifically, to force the metal-substitution reaction, the method heats the integrated-circuit assembly to 500–600° C. in a vacuum, nitrogen, argon, or other non-oxidizing atmosphere for approximately 60 minutes. Heating urges diffusion of portions of metal layer 22 into adjacent non-oxidized portions of silicon-germanium layer 18 and vice versa. The substitution temperature, that is, the annealing temperature, should be lower than 651° C., which is the eutectic (lowest melting) temperature of the ternary silver-silicon-germanium system for the reaction or substitution to occur.

Forming-gas or other hydrogen-containing atmospheres are not used at the 500–600° C. temperatures because of the tendency to saturate zirconium layer **24** with hydrogen and form zirconium hydride. However, if the process is performed below 400° C., a forming-gas atmosphere is feasible.

After the metal-substitution reaction, a small amount of germanium may remain on silver structure **22'**. Additionally, any silver that remains on the surface of layer **18** may contain zirconium silicide and silicon-germanium agglomerates. These are removed by chemical mechanical polishing or other suitable techniques.

Then, to reduce the dielectric constant of oxidized regions **18'**, germanium oxide in the oxidized regions of silicon-germanium layer **18** is removed using a wet etchant, such as distilled or deionized water. This leaves a porous oxide, insulative structure having a dielectric constant about one half that of conventional solid oxide insulators. (However, this reduction in dielectric constant depends on the original composition of silicon-germanium layer **18**, which in the preferred embodiment is 50 percent germanium.) The lowered dielectric constant ultimately reduces capacitance which, in turn, yields a faster, more efficient integrated circuit.

Further fabrication of more interconnection or metallization levels would entail repeating the steps already described and illustrated. For example, to form a second level that includes a via, the preferred embodiment, illustrated in FIG. **6A**, forms a second silicon-germanium layer **26** and then an oxidation mask **28** defining a position of the via relative conductor **22'**. FIG. **6B** shows a top view of mask **28** and its position relative conductor **22'**. After oxidizing silicon-germanium layer **26** and then removing oxidation mask **28**, the method forms a second silver layer **30** and a second zirconium layer **32** as FIG. **7** shows. The integrated-circuit assembly is then heated to force the metal-substitution reaction, thereby substituting a portion of silver layer **30** for non-oxidized region of layer **26**, and forming a substantially coplanar via **30'**.

Notably, unlike conventional techniques that require forming an insulative layer, etching holes to define the vias, and then finally filling these holes with metal to form the vias, this technique proceeds without etching holes and without filling holes. Thus, the present invention provides an “etchless” or “hole-less” method of forming vias.

After forming via **30'**, wet etching removes germanium oxide from regions **26'** to form porous low-dielectric insulation around via **30'**. The resulting integrated-circuit assembly is depicted in FIG. **8**. Subsequent metallizations would follow similarly. (However, one should take care to ensure that both the deposition and oxidation temperatures for subsequent silicon-germanium layers are lower than the substitution temperature, 651° C. in the preferred embodiment, to avoid causing substitution from underlying silver structures.)

With completion of the desired number of metallization levels, the preferred method ultimately concludes by heat-treating the integrated circuit for one to six hours at a temperature between 100 and 200° C. This heat treatment, which preferably occurs after packaging the integrated circuit in a protective housing, ensures that the metallic structures have minimum resistivity.

#### Preferred Hybrid Interconnects Incorporating Low-Capacitance Air Bridge

FIGS. **9** and **10** show two integrated-circuit assemblies which respectively combine the low-capacitance intercon-

nect structures of FIGS. **3** and **8** with low-capacitance air bridges to yield new hybrid structures. In particular, FIG. **9** shows an integrated-circuit assembly **40** embodying a coplanar hybrid interconnect system which combines two distinct types of low-capacitance interconnect structures: a first interconnect structure **42** (on the left), and a second interconnect structure **43** (on the right). Structure **42**, identical to the previously described structure of FIG. **3**, connects transistors **14a** and **14b** on substrate **12** via conductive vias (or contacts) **16a** and **16b** and aluminum conductor **22'**. For reduced capacitance, conductor **22'** is embedded in porous insulation **18''** which has a low dielectric constant. Structure **42** is preferably formed as detailed through the above-description of FIGS. **1–3**.

Interconnect structure **43** comprises an air bridge **44**, which in turn comprises an aluminum conductor **44a** and air cavity **44b**. Conductor **44a**, which is substantially coplanar to conductor **22'**, electrically connects vias **16c** and **16d** and thus electrically connects corresponding transistors **14c** and **14d**. The presence of air cavity **44b**, which has a unity or near-unity dielectric constant lower than even that of porous insulation **18''**, minimizes line capacitance between these transistors.

Air bridge **44** is built conventionally by forming conductor **44a** on an underlying, sacrificial support structure (not shown) and then removing the support structure to leave air cavity **44b**. In conventional processing, the sacrificial support structure consists of photoresist or other material which can be easily dissolved or etched away. For further details on this known process, see U.S. Pat. No. 5,510,645 entitled Semiconductor Structure Having an Air Bridge and Method of Forming the Semiconductor Structure and U.S. Pat. No. 5,324,684 entitled Method of Forming Semiconductor Structure Having an Air Bridge, both issued to Fitch et al. and both incorporated herein by reference.

Fabrication of air bridge **44** may occur before, after, or concurrent with the fabrication of structure **42**. For example, vias **16c** and **16d** may be formed simultaneous with vias **16a** and **16b** or started during the metal substitution reaction that forms conductor **22'**. However, one should take care to avoid exceeding the substitution temperature; otherwise undesired substitution of silver may occur.

FIG. **10** shows an integrated-circuit assembly **50** embodying a two-level (non-coplanar) hybrid interconnect system which, like assembly **40**, combines two types of low-capacitance interconnect structures. Assembly **50** includes an air bridge **54** which connects integrated-circuit assemblies **52** and **53**. Assemblies **52** and **53**, both of which are structurally identical to the previously described assembly of FIG. **8**, embed aluminum conductors **22'** in porous insulation **26''**. Conductors **22''** are connected to vias **30'**, which are electrically connected via air bridge **54**.

Air bridge **54** comprises aluminum conductor **54a** and air cavity **54b**. Conductor **54a**, which occupies a plane above that of conductors **22'**, electrically connects vias **30'** and thus electrically connects transistors **14a** and **14b** to transistors **14c** and **14d**. Air bridge **54** as well as assemblies **52** and **53** are fabricated according the methods described above.

#### Conclusion

The present invention overcomes at least three significant shortcomings of conventional interconnection techniques. First, unlike previous techniques that require digging trenches in an insulative layer before metallization to ensure a coplanar metallic structure, the invention provides a self-planarizing or self-trenching metallization process, which



directly substitutes metal for select portions of an insulative layer, thereby skipping the time-consuming trench-digging step. Relatedly, the invention forms vias without the conventional steps of etching and filling holes with metal.

Second, unlike conventional techniques which are limited to forming interconnects from aluminum, the invention forms interconnects from silver, which has 40-percent lower electrical resistance and at least 100-percent higher electromigration resistance. Thus, the invention yields integrated circuits with superior efficiency and reliability.

Third, unlike conventional techniques that bury aluminum wiring in solid high-capacitive insulation, the invention places silver wiring in porous low-capacitive insulation which ultimately yields faster integrated circuits. In short, the invention not only eliminates the time-consuming, trench-digging step, but also yields integrated circuits that operate with superior speed, reliability, and economy.

The embodiments described above are intended only to illustrate and teach one or more ways of practicing or implementing the present invention, not to restrict its breadth or scope. The actual scope of the invention, which embraces all ways of practicing or implementing the invention, is defined only by the following claims and their equivalents.

What is claimed is:

**1.** A method of making a silver structure, the method comprising:

forming a first layer including silicon and germanium;

oxidizing a region of the first layer to define an oxidized region and a non-oxidized region; and

substituting silver for at least a portion of the non-oxidized region to form the silver structure.

**2.** The method of claim 1, wherein forming the first layer comprises depositing a silicon alloy having between 10 and 60 percent germanium.

**3.** The method of claim 1, wherein forming the first layer comprises ion-vapor deposition or photo-enhanced vapor deposition.

**4.** The method of claim 1, wherein oxidizing a region of the first layer comprises forming a mask structure on the first layer.

**5.** The method of claim 1, wherein substituting silver for at least a portion of the non-oxidized region occurs within a non-oxidizing atmosphere or a vacuum.

**6.** The method of claim 1, wherein substituting silver for at least a portion of the non-oxidized region comprises:

forming a layer including silver on the non-oxidized region of the first layer; and

heating the layer including silver.

**7.** The method of claim 6, wherein heating the layer including silver comprises:

maintaining its temperature at about 300° C. for about an hour.

**8.** The method of claim 6, further comprising

forming a layer including titanium, zirconium, or hafnium on the layer including silver.

**9.** The method of claim 1, further comprising reducing the dielectric constant of the oxidized region.

**10.** The method of claim 9, wherein reducing the dielectric constant of the oxidized region comprises removing germanium oxide from the oxidized region.

**11.** The method of claim 1, further comprising:

heating the silver structure to a predetermined temperature for a predetermined amount of time to reduce its electrical resistivity.

**12.** The method of claim 1, wherein oxidizing a region of the first layer occurs before substituting silver for at least a portion of the non-oxidized region.

**13.** A method of making coplanar silver and insulative structures, the method comprising:

forming a layer consisting essentially of silicon and germanium;

forming a mask structure on the layer to define a desired metallic region and a desired insulative region;

oxidizing the desired insulative region to form an insulative structure comprising a germanium oxide;

removing the mask structure; and

substituting silver for at least a portion of the desired metallic region, thereby forming a silver structure substantially coplanar with the insulative structure.

**14.** The method of claim 13, wherein forming the layer comprises depositing a silicon alloy having between 10 and 60 percent germanium.

**15.** The method of claim 13, wherein forming the layer comprises ion-vapor deposition or photo-enhanced vapor deposition.

**16.** The method of claim 13, wherein substituting silver for at least a portion of the desired metallic region comprises:

forming a metal layer consisting essentially of silver on the desired metallic region of the layer; and

heating the metal layer.

**17.** The method of claim 16, wherein heating the metal layer comprises:

increasing its temperature to about 300° C.

**18.** The method of claim 16, wherein heating the metal layer occurs in a nitrogen, forming gas, or other non-oxidizing atmosphere.

**19.** The method of claim 16, further comprising

forming a layer including titanium, zirconium, or hafnium on the metal layer.

**20.** The method of claim 13, further comprising:

reducing the dielectric constant of the insulative structure by removing at least a portion of the germanium oxide.

**21.** The method of claim 13, further comprising:

heating the silver structure to a predetermined temperature for a predetermined amount of time to reduce its electrical resistivity.

**22.** A method of embedding a silver structure within a structure having a reduced-dielectric constant, the method comprising:

forming a first structure comprising an extractable material and a metal-substitutable material;

substituting silver for at least some of the metal-substitutable material to form the silver structure; and

extracting the extractable material from the first structure to form the structure having the reduced-dielectric constant.

**23.** A method of electrically connecting first and second integrated devices in an integrated-circuit assembly, the method comprising:

forming a first layer including first and second conductive members contacting the respective first and second integrated devices;

forming a silicon-germanium layer contacting the first and second conductive members, thereby defining first and second desired metallic regions of the silicon-germanium layer; and

substituting silver for at least a portion of the silicon-germanium layer that extends from the first desired

**9**

metallic region to the second desired metallic region, thereby forming a silver conductor that electrically connects the first and second integrated devices.

**24.** A via-making method that does not require etching and filling a hole in an insulative layer, the method comprising:

forming a silicon-germanium layer on an integrated-circuit assembly;

oxidizing a region of the silicon-germanium layer to define a desired via region; and

substituting silver for at least a portion of the desired via region.

**25.** A method of making an integrated circuit, comprising:

forming an air bridge having an air-bridge conductor; and forming a silicon-germanium layer;

oxidizing a region of the silicon-germanium layer to define a desired insulative region and a desired conductive region; and

substituting silver for at least a portion of the desired conductive region to form a second conductive member electrically coupled to the air-bridge conductor.

**10**

**26.** The method of claim **25** wherein forming the air bridge comprises:

forming a support structure;

forming the conductive member on the support structure; and

removing the support structure.

**27.** The method of claim **25** wherein forming the air bridge occurs after substituting silver for at least the portion of the desired conductive region.

**28.** A method of embedding a silver structure within a structure having a reduced-dielectric constant, the method comprising:

forming a first structure comprising an extractable material and a metal-substitutable material;

substituting silver for at least some of the metal-substitutable material to form the silver structure; and

extracting the extractable material from the first structure to form the structure having the reduced-dielectric constant, after substituting silver for at least some of the metal-substitutable material.

\* \* \* \* \*